

Gas Grade Recommendations

Analytical Method/ Detector	Carrier and Support Gases	Type of Analysis Impurity Considerations
Gas Chromatography		
TCD (Thermal Conductivity Detector)	N ₂	Universal Detector Atmospheric contaminants can oxidize the detector filament giving rise to negative peaks and reduced sensitivity.
	He	
	H ₂	
	Ar	
FID (Flame Ionization Detector) Carriers Combustion Gases	He	Organic Compounds Hydrocarbons in carrier and fuel gases can give rise to baseline noise and reduced detector sensitivity. Oxygen and water cause column deterioration and affect retention time on critical separations.
	N ₂	
	Ar	
	H ₂	
	40% H ₂ in He	
	40% H ₂ in N ₂	
	Air	
ECD (Electron Capture Detector)	He	Halogenated Compounds Detector response and column life are reduced by oxygen and water. Hydrocarbons and halocarbons can produce baseline noise, negative peaks and plumbing contamination.
	N ₂	
	5% CH ₄ in Ar-ECD (P-5)	
	10% CH ₄ in Ar-ECD (P-10)	
HID (Helium Ionization Detector)	He	Universal Detector Atmospheric impurities can cause baseline noise signal polarity and reduced detector stability and sensitivity.
	He Purge	
FPD (Flame Photometric Detector)	He	Sulfur or Phosphorous Compounds Organics can yield baseline noise and carbon dioxide can suppress detector response.
	N ₂	
	H ₂	
	Air	
PID (Photo Ionization Detector)	He	Selective Detector Dependent on UV Source Organics can yield baseline noise and carbon dioxide can suppress detector response.
	N ₂	
GC/MS (Mass Spectrometer)	He	All Compounds Organics can yield baseline noise and carbon dioxide can suppress detector response.
	N ₂	
	Ar	
DID (Discharge Ionization Detector)	He	Universal Detector Atmospheric impurities can cause baseline noise signal polarity and reduced detector stability and sensitivity.
	He Purge	
USD (Ultrasonic Detector)	Ar	Universal Detector Atmospheric impurities can cause baseline noise signal polarity and reduced detector stability and sensitivity.
	He	

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Optical Spectrometry – Adsorption			
NDR(Non-dispersive Infrared)	Air		Polyatomic and Heteroatomic Compounds
	N ₂		
IR (Dispersive Infrared) -FTIR (Fourier Transform Infrared) -FG/GFC (Correlation)	Ar		Polyatomic and Heteroatomic Compounds During matrix isolation, oxygen can oxidize a sample. Moisture interferes with IR spectra. Impurities coinciding with quantified peaks can cause inaccuracies.
	N ₂		
AA (Atomic Absorption) Combustion Gases	C ₂ H ₂		Elemental Analysis Impurities can cause the flame to discolor or burn unevenly. Furnace atmospheres require low oxygen and moisture levels to maintain instrument sensitivity.
	n-C ₄ H ₁₀		
	H ₂		
	N ₂ O		
	Air		
	Ar (Flameless)		
	N ₂		
NMR (Nuclear Magnetic Resonance)	LHe		Analysis of Molecular Structure
	LN ₂		

Legend

- AA – Atomic Absorption
- UH– Ultra High Purity
- RS – Research/Chromatographic
- Z – Zero
- UZ – Ultra Zero
- IS – Instrument
- OF – Oxygen Free
- TG – Trace Analytical
- EC – Electron Capture Detector

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Analytical Method/ Detector	Carrier and Support Gases		Type of Analysis Impurity Considerations
Optical Spectrometry – Emission			
Atomic Emission - ICP (Inductive Coupled Plasma)	Ar		Elemental Analysis
	LAr		
	N ₂		
	H ₂		
Arc or Spark Emission	Ar		Elemental Analysis
	H ₂		
	5% Ar in H ₂		
Chemiluminescence	Air		NO-NO ₂ -NOX Hydrides and O ₃
	N ₂		
	O ₂		
Fluorescence UV	Air		SO ₂ -H ₂ S-Organic Compounds
	N ₂		
XRF (Fluorescence X)	10% CH ₄ in Ar		Elemental Analysis
	1.3% n-C ₄ H ₁₀ in He		
	0.95% i-C ₄ H ₁₀ in He		
	LN ₂		
Mass Spectrometry			
MS (Under Vacuum)			All Compounds
	N ₂		
	He		

Legend

UH – Ultra High Purity

RS – Research

IC – Inductive Coupled Plasma

CE – Continuous Emissions Monitoring
(See Pure Gas Section for correct
definition)

TG – Trace Analytical

Z – Zero

VC – Volatile Organic Compound Free

UZ – Ultra Zero

Gas Grade Recommendations

Analytical Method/ Detector	Carrier and Support Gases		Type of Analysis Impurity Considerations
Others			
Nuclear Counter	10% CH ₄ in Ar		Measurement of Radioactivity
	1.3% n-C ₄ H ₁₀ in He		
	0.95% i-C ₄ H ₁₀ in He		
Hydrometer	Air		Moisture in All Gas
	N ₂		
Paramagnetic Analyzer	N ₂		Oxygen in All Gas
	O ₂ in N ₂		
Carbon and Sulfur in Steel	Ar		Analysis of Carbon, Sulfur and Gases (N ₂ -H ₂ -O ₂) in Steel
	He		
	N ₂		
	O ₂		

Instrumentation Mixture Summary

Product Description	Mixture Application
40% H ₂ in He (FID Fuel) (THC < 0.5 ppm)	Fuel Gas for GC-FID
40% H ₂ in He UHP (FID Fuel) (THC < 0.1 ppm)	Fuel Gas for GC-FID
40% H ₂ in N ₂ (FID Fuel) (THC < 0.5 ppm)	Fuel Gas for GC-FID
40% H ₂ in N ₂ UH (FID Fuel) (THC < 0.1 ppm)	Fuel Gas for GC-FID
5% CH ₄ in Argon-ECD (P-5)	Make-up Gas for GC-ECD
10% CH ₄ in Argon-ECD (P-10)	Make-up Gas for GC-ECD
10% CH ₄ in Ar (P-10)	XRF (Fluorescence X)
1.3% n-C ₄ H ₁₀ in He	XRF (Fluorescence X)
0.95% i-C ₄ H ₁₀ in He	XRF (Fluorescence X)
5% CH ₄ in Ar (P-5)	Carrier Gas for Proportional Counters
10% CH ₄ in Ar (P-10)	Carrier Gas for Proportional Counters
1.3% n-C ₄ H ₁₀ in He	Quench Gas
0.95% i-C ₄ H ₁₀ in He	Carrier Gas for Geiger-Muller
5% Ar in H ₂	Spark Emission